

SI Dimensions of Physical Quantities listed by Category

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ALPHABETIC LIST | SI Units | Footnotes

PHYSICS and MATH constants

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Quantity	Dimension	Alternatives	Root definition and Notes
Basic SI quantities			
Length	m	m	meter
Mass	kg	kg	kilogram
Time	s	s	second
Current, electric	A	A	ampere
Temperature	K	K	kelvin
Quantity of substance	mol	mol	mole
Luminosity Luminous intensity	cd	cd	candle
<i>Pseudo-dimensional quantities:</i>			
Plane angle	1	rad	radian
Solid angle	1	sr	steradian
Universal dimensionless quantities			
Count of events Number of instances	1		This covers all kinds of enumerations
Probability of an event	1		Real number in a dimensionless interval [0,1]
Ratio of commensurable quantities	1		Q1/Q2, with Q1 and Q2 having the same dimension
Relative variation	1		$\Delta Q/Q$, for any quantity Q
Logarithmic ratio $\log_b(A/A')$ in any base b	1		Applicable to any ratio of commensurable quantities
Logarithmic scale differential Relative differential	1		$d\{\ln(Q)\} = dQ/Q$, for any quantity Q
<i>Pseudo-dimensional quantities:</i>			
Phase Phase angle	1	rad	φ typically in $\exp(i(\omega t + \varphi))$
Logarithmic ratio $\text{Log}(P/P')/10$	1	dB	decibel . Uses base-10 logarithm. Applies to power P
Logarithmic ratio $\text{Log}(X/X')/20$	1	dB	decibel . Uses base-10 logarithm. Applies to amplitudes X
Gain or Loss of a device	1	usually in dB	[Output]/[Input], provided they are commensurable quantities
Attenuation Amplification (generic)	1	usually in dB	[Quantity(p)]/[Quantity(p')], with p being some parameter
Logarithmic ratio $\ln(A/A')$	1	Np	Neper . Uses natural logarithm
Logarithmic scale probability density	1	1/Np	[Probability]/[Natural-logarithmic ratio]
Operators			
Derivative with respect to time	s^{-1}		$d/dt, \partial/\partial t$
Derivative with respect to a length	m^{-1}		$d/dr, \partial/\partial r, r = x y z$
Nabla (∇) div grad rot curl	m^{-1}		Any derivative-like construct with respect to a distance
Laplace operator Laplacian	m^{-2}		$\nabla^2 = \partial^2/\partial x^2 + \partial^2/\partial y^2 + \partial^2/\partial z^2$
D'Alembert operator D'Alembertian	m^{-2}		$(1/c^2)\partial^2/\partial t^2 - \partial^2/\partial x^2 - \partial^2/\partial y^2 - \partial^2/\partial z^2$
Multiple derivatives with respect to time	s^{-p}		$d^p/dt^p, \partial^p/\partial t^p$; for $p = 1, 2, 3, \dots$
Multiple derivatives with respect to a length	m^{-p}		$d^p/dr^p, \partial^p/\partial r^p$; for $p = 1, 2, 3, \dots, r = x y z$
Quantities related only to time			
Time Duration	s	s	second
Half life	s		of a non-conservative / decaying quantity
Settling time	s	typically dB/s	Used to describe transient phenomena

Relaxation time	s		Used for returns to equilibria
Activity Frequency of events	s ⁻¹		[Counts]/[Time]
Count rate Expectation frequency	s ⁻¹		[Counts]/[Time]
Relative growth rate	s ⁻¹		[Relative variation]/[Time]
Relative evolution rate Log-scale evolution rate	s ⁻¹		d{ln(Q)}/dt = (dQ/dt)/Q
Settling rate	s ⁻¹	typically dB/s	[Ratio]/[ΔTime]. Used for transient phenomena
Relaxation rate	s ⁻¹		1/[Relaxation time]
Frequency of waves	s ⁻¹	Hz	hertz
Phase drift rate	s ⁻¹	rad.s ⁻¹	[Phase angle]/[Time]
Angular velocity / speed	s ⁻¹	rad.s ⁻¹	[Plane angle]/[Time]
Frequency drift rate	s ⁻²	Hz.s ⁻¹	[ΔFrequency]/[Time]. Applicable to waves
Angular acceleration / deceleration	s ⁻²	rad.s ⁻²	[ΔAngularVelocity]/[Time]

Quantities related only to space

Position vector	m		in all Euclidean n-dimensional spaces
Length Distance	m	m	meter
Perimeter Circumference Radius	m		
Thickness	m		usually referred to planar structures
Wavelength	m		[Wave velocity]/[Frequency]
Wavenumber	m ⁻¹		[Number of waves]/[Distance]
K-space vector Reciprocal space position	m ⁻¹		
Curvature radius	m		of a line in plane/space or surface in space
Curvature	m ⁻¹		1/[Curvature radius]
Convergence	m ⁻¹	dioptre	used in optics, but not only ..
Attenuation / amplification over a distance	m ⁻¹	dB/m	[Attenuation]/[Distance]. Mostly in acoustic and electronics
Extinction coefficient	m ⁻¹	dB/m	[Ratio]/m. Used mostly for radiation
Propagation / transmission loss	m ⁻¹	dB/m	[Ratio]/m. Generic, usable for any quantity
Area Cross section	m ²		[Distance]*[Distance]
Surface element Surface area	m ²		[Distance]*[Distance]. Applicable to 3D bodies
Volume element Volume	m ³		[Area]*[Distance]

Propagation through space and time

Velocity Speed	m.s ⁻¹		[Distance]/[Time]
Acceleration Deceleration	m.s ⁻²		[ΔVelocity]/[ΔTime]
Drift speed	m.s ⁻¹		Steady-state speed of an object
Surface / area growth rate	m ² .s ⁻¹		[ΔArea]/[Time]
Volume growth rate	m ³ .s ⁻¹		[ΔVolume]/[Time]. For example, of a crystal
Volume flow	m ³ .s ⁻¹		[Volume]/[Time]. For example, through a device

Matter distribution and transport

Particle density	m ⁻³		[Count]/[Volume]. Obsolete: number density
Mass	kg	kg	kilogram
Mass production rate	kg.s ⁻¹		[ΔMass]/[Time]
Mass density Specific density	kg.m ⁻³		[Mass]/[Volume]
Mass density gradient Specific density gradient	kg.m ⁻⁴		[Mass density]/[Distance]
Specific volume	m ³ .kg ⁻¹		[Volume]/[Mass]
Concentration ratio by volume	1	Dimensionless	[Partial volume]/[Total volume]
Concentration ratio by mass	1	Dimensionless	[Partial mass]/[Total mass]. Not <i>by weight</i> . obsolete)
Mass flow (total)	kg.s ⁻¹		[ΔMass]/[Time]. For example, through a device
Diffusion coefficient	m ² .s ⁻¹		[Distance ²]/[Time]

Molar distribution and transport quantities:

Particle count, molar	mol ⁻¹		[Count]/[Mol]. For example, the Avogadro constant
Molar production rate	mol.s ⁻¹		[ΔQuantity]/[Time]
Molar mass	kg.mol ⁻¹		[Mass]/[Quantity]
Molar volume	m ³ .mol ⁻¹		[Volume]/[Quantity]
Molar density Density of substance	m ⁻³ .mol		[Quantity]/[Volume]
Molarity Concentration	m ⁻³ .mol		[Quantity]/[Volume]. Same as molar density

Molarity gradient Concentration gradient	$m^{-4}.mol$		[Molarity]/[Distance]
Molar concentration ratio	1	Dimensionless	[Partial quantity]/[Total quantity]
Molality (intended as concentration)	$kg^{-1}.mol$	mol/kg	[Quantity]/[Mass]. Obsolete
Katalytic activity	$mol.s^{-1}$	katal	[ΔQuantity]/[Time]

Mechanics and hydrodynamics

Force	$kg.m.s^{-2}$	N	newton . [Mass]*[Acceleration]
Moment of motion	$kg.m.s^{-1}$		[Mass]*[Velocity], [Mass flow]*[Distance]
Impulse	$kg.m.s^{-1}$		[ΔMoment of motion], [Force]*[ΔTime], [Mass]*[ΔVelocity]
Moment of force Torque	$kg.m^2.s^{-2}$	N.m	[Force]*[Distance]. Like energy
Couple	$kg.m^2.s^{-2}$	N.m	2*[Force]*[Distance] for two non-aligned opposing forces
Pressure	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	pascal . [Force]/[Area]
Pressure gradient	$kg.m^{-2}.s^{-2}$	$N.m^{-3}$, Pa/m	[Pressure]/[Distance]
Energy Lagrangian Hamiltonian	$kg.m^2.s^{-2}$	N.m, J	joule . [Force]*[Distance], [Power]*[Time]
Specific energy	$m^2.s^{-2}$	$J.kg^{-1}$	[Energy]/[Mass]
Energy density	$kg.m^{-1}.s^{-2}$	$J.m^{-3}$	[Energy]/[Volume]
Power Energy flux	$kg.m^2.s^{-3}$	$J.s^{-1}$, W	watt . [ΔEnergy]/[ΔTime]
Action	$kg.m^2.s^{-1}$	J.s	[Energy]*[Time], [Moment of motion]*[Distance]
Angular moment of inertia	$kg.m^2$		[Mass]*[Distance ²]
Angular moment of motion	$kg.m^2.s^{-1}$	J.s	[Moment of motion]*[Distance]
Circulation	$m^2.s^{-1}$	$J.s.kg^{-1}$	[Angular moment]/[Mass], [Velocity]*[Loop length]
Spin	1	Dimensionless	of a quantum particle
Stress Tension Compression	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa (pascal)	[Force]/[Area]. ... same as pressure
Compressive strength	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	[Force]/[Area]. Like pressure
Strain (mechanical)	1	Dimensionless	[ΔLength]/[Length] Relative deformation
Friction	$kg.m.s^{-2}$	N	Tangential force between two moving surfaces
Traction	$kg.m.s^{-2}$	N	Maximum tangential force before slipping
Velocity, superficial	$m.s^{-1}$	m/s	In porous media ; as if the space was filled only by the fluid
Velocity, advection	$m.s^{-1}$	m/s	In porous media ; actual progress along pressure gradient
Wave function for N particles (quantum)	$m^{-3N/2}$	tentative	$ \psi ^2 dt^N$ is a dimensionless probability element.

Mechanical and hydrodynamic properties of matter

Compressibility Modulus of compression	$kg^{-1}.m.s^2$	Pa^{-1}	[Pressure]/([ΔVolume]/[Volume]). Inverse of bulk modulus
Bulk modulus	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	([ΔVolume]/[Volume])/[Pressure]. Inverse of compressibility
Young modulus	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	[Stress]/[Strain]. Like shear modulus
Shear modulus Modulus of rigidity	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	[Stress]/[Strain]. Same dimension as Young modulus
Poisson's ratio	1	Dimensionless	[Transversal striction]/[Longitudinal elongation]
Impact Notch resistance	$kg.s^{-2}$	$J.m^{-2}$	[Energy]/[Area]
Hardness Tensile strength	$kg.m^{-1}.s^{-2}$	$N.m^{-2}$, Pa	[Force]/[Area]. Like pressure
Stiffness (linear)	$kg.s^{-2}$	$N.m^{-1}$	[Force]/[Displacement]. ... of a structure
Stiffness (rotational)	$kg.m^2.s^{-2}.rad^{-1}$	$N.m.rad^{-1}$	[Moment of force]/[Angle]. ... of a structure
Friction coefficient	1	Dimensionless	[Tangential force]/[Normal force]
Traction coefficient	1	Dimensionless	[Traction]/[Weight]
Self-diffusion coefficient	$m^2.s^{-1}$		[Distance ²]/[Time]
Surface tension	$kg.s^{-2}$	N/m	[Force]/[Length]. Same as surface energy
Surface energy	$kg.s^{-2}$	J/m^2	[Energy]/[Area]. Same as surface tension
Viscosity, dynamic	$kg.m^{-1}.s^{-1}$	Pa.s	([Force]/[Area])/[ΔVelocity]
Viscosity, kinematic	$m^2.s^{-1}$		[Dynamic viscosity]/[Density]
Reynolds number	1	Dimensionless	[Velocity]*[length]/[Kinematic viscosity]
Critical angle of repose	rad	or degree	Steepest angle of a slope before a slide
Porosity, volume	1	Dimensionless	[Volume of pores]/[Total volume], in porous media
Porosity, superficial	1	Dimensionless	[Void cross section]/[Total cross section], in porous media
Permeability, hydraulic	m^2	1 darcy = $10^{-12} m^2$	[Velocity]*[Viscosity]/[Pressure gradient], in porous media
Conductivity, hydraulic	$m.s^{-1}$	m/s	Used for porous media
Specific acoustic impedance / resistance / reactance	$kg.m^{-2}.s^{-1}$	Pa.s/m, reyl	[ΔPressure]*[Velocity], intensive property
Specific acoustic conductance / susceptance	$kg^{-1}.m^2.s$	$reyl^{-1}$	Inverse of specific acoustic impedance
Acoustic impedance / resistance / reactance	$kg.m^{-4}.s^{-1}$	Pa.s/m ³ , $reyl/m^2$	[ΔPressure]/[Volume flow rate], extensive property

Thermodynamics

Temperature	K	K	kelvin
Temperature gradient Thermal gradient	$K.m^{-1}$		$[\Delta\text{Temperature}]/[\text{Distance}]$
Heat Internal energy Enthalpy	$kg.m^2.s^{-2}$	J	Same as energy
Specific heat internal energy enthalpy	$m^2.s^{-2}$	$J.kg^{-1}$	$[\text{Heat}]/[\text{Mass}]$
Heat capacity	$kg.m^2.s^{-2}.K^{-1}$	$J.K^{-1}$	$[\Delta\text{Heat}]/[\Delta\text{Temperature}]$
Heat flux	$kg.m^2.s^{-3}$	J.s, W	$[\Delta\text{Heat}]/[\Delta\text{Time}]$. Same as power
Heat flux density Irradiance	$kg.s^{-3}$	$W.m^{-2}$	$[\text{Heat flux}]/[\text{Area}]$
Entropy	$kg.m^2.s^{-2}.K^{-1}$	$J.K^{-1}$	$[\Delta\text{Heat}]/[\text{Temperature}]$
Specific entropy	$m^2.s^{-2}.K^{-1}$	$J.K^{-1}.kg^{-1}$	$[\text{Entropy}]/[\text{Mass}]$
Free energy Free enthalpy	$kg.m^2.s^{-2}$	J	Helmholtz Gibbs functions , respectively
Specific free energy free enthalpy	$m^2.s^{-2}$	$J.kg^{-1}$	$[\text{Energy}]/[\text{Mass}]$. Also specific Helmholtz Gibbs functions
Molar thermodynamical quantities:			
Molar heat internal energy enthalpy	$kg.m^2.s^{-2}.mol^{-1}$	$J.mol^{-1}$	$[\text{Heat}]/[\text{Quantity}]$
Molar energy	$kg.m^2.s^{-2}.mol^{-1}$	$J.mol^{-1}$	$[\text{Energy}]/[\text{Quantity}]$
Molar entropy	$kg.m^2.s^{-2}.K^{-1}.mol^{-1}$	$J.K^{-1}.mol^{-1}$	$[\text{Entropy}]/[\text{Quantity}]$
Molar free energy free enthalpy	$kg.m^2.s^{-2}.mol^{-1}$	$J.mol^{-1}$	$[\text{Energy}]/[\text{Quantity}]$. Molar versions of the above

Thermodynamic and thermal properties of matter

Thermal expansion coefficient	K^{-1}		$([\Delta\text{Length}]/[\text{Length}])/[\text{Temperature}]$
Heat capacity, specific	$m^2.s^{-2}.K^{-1}$	$J.K^{-1}.kg^{-1}$	$[\text{Heat capacity}]/[\text{Mass}]$
Heat capacity, molar	$kg.m^2.s^{-2}.K^{-1}.mol^{-1}$	$J.K^{-1}.mol^{-1}$	$[\text{Heat capacity}]/[\text{Quantity}]$
Heat of fusion evaporation, specific	$m^2.s^{-2}$	$J.kg^{-1}$	$[\text{Energy}]/[\text{Mass}]$
Heat of fusion evaporation, molar	$kg.m^2.s^{-2}.mol^{-1}$	$J.mol^{-1}$	$[\text{Energy}]/[\text{Quantity}]$
Heat conductivity	$kg.m.s^{-3}.K^{-1}$	$W.m^{-1}.K^{-1}$	$[\text{Heat flux}]/([\text{Distance}]*[\Delta\text{Temperature}])$
Thermal diffusivity	$m^2.s^{-1}$		$([\partial\text{Temp}]/[\partial\text{Time}])/[\nabla^2\text{Temp}]$.
Prandtl number	1	Dimensionless	$[\text{Kinematic viscosity}]/[\text{Thermal diffusivity}]$
Joule-Thomson coefficient	$kg^{-1}.m.s^2.K$	$K.Pa^{-1}$	$[\Delta\text{Temperature}]/[\Delta\text{Pressure}]$
Pi coefficient, molar	$kg.m^{-1}.s^{-2}.mol^{-1}$	$J.m^{-3}$	$[\Delta\text{InternalEnergy}]/[\Delta\text{Volume}]$
Chemical potential, molar	$kg.m^2.s^{-2}.mol^{-1}$	$J.mol^{-1}$	$[\Delta\text{InternalEnergy}]/[\Delta\text{Quantity}]$
Softening point	K		Temperature at which hardness drops below a level
Annealing point	K		Temperature at which viscosity drops below 10^{12} Pa.s
Strain point	K		Temperature at which viscosity drops below $10^{13.5}$ Pa.s
Flash point	K		Temperature at which vapour can be kept burning
Fire point	K		Temperature at which ignited vapour keeps burning

Thermal properties of devices

Thermal resistance	$kg^{-1}.m^{-2}.s^3.K$	K/W	$[\Delta T]/[\text{Power}]$.
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Electromagnetism

Charge, electric	s.A	C	coulomb . $[\text{Current}]*[\text{Time}]$
Charge density	$m^{-3}.s.A$	$C.m^{-3}$	$[\text{Charge}]/[\text{Volume}]$
Current, electric	A	A	ampere . $[\text{Charge}]/[\text{Time}]$
Current density Current intensity	$m^{-2}.A$		$[\text{Current}]/[\text{Area}]$
Specific charge Charge/mass ratio	$kg^{-1}.s.A$	$C.kg^{-1}$	$[\text{Charge}]/[\text{Mass}]$
Molar charge	$s.A.mol^{-1}$	$C.mol^{-1}$	$[\text{Charge}]/[\text{Quantity}]$
Quantum charge	1	Dimensionless	$[\text{Charge}]/[\text{Elementary charge quantum}]$
Surface density of charge	$m^{-2}.s.A$	$C.m^{-2}$	$[\text{Charge}]/[\text{Area}]$
Potential, electric	$kg.m^2.s^{-3}.A^{-1}$	$W.A^{-1}, J.C^{-1}, C.F^{-1}, V$	volt . $[\text{Power}]/[\text{Current}]$, $[\text{Energy}]/[\text{Charge}]$
Electric dipole moment	$m.s.A$	$C.m$	$[\text{Charge}]*[\text{Distance}]$
Electric quadrupole moment	$m^2.s.A$	$C.m^2$	$[\text{Electric dipole}]*[\text{Distance}]$, $[\text{Electric charge}]*[\text{Distance}^2]$
Electric field strength Electric intensity	$kg.m.s^{-3}.A^{-1}$	$V.m^{-1}$	$[\Delta\text{Potential}]/[\text{Distance}]$
Electric field gradient	$kg.s^{-3}.A^{-1}$	$V.m^{-2}$	$[\Delta E].\text{field strength}]/[\text{Distance}]$
Electric flux density Electric induction	$m^{-2}.s.A$	$C.m^{-2}$	$[\text{Charge}]/[\text{Area}]$
Electric polarization Electric displacement	$m^{-2}.s.A$	$C.m^{-2}$	$[\text{Charge}]/[\text{Area}]$. Same as electric flux density
Magnetic field strength Magnetic intensity	$m^{-1}.A$		$[\text{Current}]/[\text{Distance}]$
Magnetic flux	$kg.m^2.s^{-2}.A^{-1}$	$V.s, W.s.A^{-1}, Wb$	weber . $[\Delta\text{Potential}]*[\text{Time}]$, $[\text{Power}]/[d\text{Current}/dt]$

Magnetic flux density Magnetic induction	$\text{kg}\cdot\text{s}^{-2}\cdot\text{A}^{-1}$	$\text{Wb}\cdot\text{m}^{-2}$, T	tesla . [Mag.flux]/[Area]
Magnetic vector potential	$\text{kg}\cdot\text{m}\cdot\text{s}^{-2}\cdot\text{A}^{-1}$	$\text{m}^{-1}\cdot\text{s}\cdot\text{V}$, $\text{m}\cdot\text{T}$	[Mag.flux density]*[Distance], [El.field strength]*[Time]
Magnetization	$\text{m}^{-1}\cdot\text{A}$		[Magnetic moment]/[Volume]. Like magnetic field strength
Magnetic charge (bound)	$\text{m}^{-2}\cdot\text{A}$		$-\nabla\cdot[\text{Magnetization}]$, -Divergence of magnetization
Poynting vector	$\text{kg}\cdot\text{s}^{-3}$	$\text{W}\cdot\text{m}^{-2}$	[El.field strength]/[Mag.field strength]. Same as irradiance
Magnetic field gradient	$\text{kg}\cdot\text{m}^{-1}\cdot\text{s}^{-2}\cdot\text{A}^{-1}$	$\text{T}\cdot\text{m}^{-1}$	$[\Delta\text{Magnetic flux density}]/[\text{Distance}]$
Magnetic dipole moment	$\text{m}^2\cdot\text{A}$	$\text{J}\cdot\text{T}^{-1}$	[Current]*[Area]. Same as magnetic moment
Magnetic quadrupole moment	$\text{m}^3\cdot\text{A}$	$\text{m}\cdot\text{J}\cdot\text{T}^{-1}$	[Magnetic dipole]*[Distance]
Gyromagnetic ratio	$\text{kg}^{-1}\cdot\text{s}\cdot\text{A}$	$\text{Hz}\cdot\text{T}^{-1}$	[Mag.moment]/[Angular moment of motion]
Magnetogyric ratio	$\text{kg}\cdot\text{s}^{-1}\cdot\text{A}^{-1}$	$\text{T}\cdot\text{Hz}^{-1}$	[Angular moment of motion]/[Mag.moment]
Relativistic four-current (J^α)	$\text{m}^{-2}\cdot\text{A}$		Like current density and [Charge]*[c]
Relativistic four-potential (A^α)	$\text{kg}\cdot\text{m}\cdot\text{s}^{-2}\cdot\text{A}^{-1}$	$\text{m}^{-1}\cdot\text{s}\cdot\text{V}$, $\text{m}\cdot\text{T}$	Like magnetic vector potential and [El.potential]/[c]
Relativistic electromagnetic field tensor ($F^{\mu\nu}$)	$\text{kg}\cdot\text{s}^{-2}\cdot\text{A}^{-1}$	T	Like magnetic flux density
Relativistic displacement four-tensor ($D^{\mu\nu}$)	$\text{m}^{-1}\cdot\text{A}$		Like magnetic intensity

Electromagnetic properties of matter

Resistivity	$\text{kg}\cdot\text{m}^3\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	$\Omega\cdot\text{m}$	[Resistance]*[Length]/[Area]
Conductivity	$\text{kg}^{-1}\cdot\text{m}^{-3}\cdot\text{s}^3\cdot\text{A}^2$	$\text{S}\cdot\text{m}^{-1}$	1/[Resistivity]
Permittivity, electric	$\text{kg}^{-1}\cdot\text{m}^{-3}\cdot\text{s}^4\cdot\text{A}^2$	$\text{F}\cdot\text{m}^{-1}$	[El.flux density]/[El.field strength]
Dielectric constant Relative permittivity	1	Dimensionless	[Permittivity]/[Permittivity of vacuum]
Permeability, magnetic	$\text{kg}\cdot\text{m}\cdot\text{s}^{-2}\cdot\text{A}^{-2}$	$\text{N}\cdot\text{A}^{-2}$, $\text{H}\cdot\text{m}^{-1}$	[Mag.flux density]/[Mag.field strength]
Reluctance, magnetic	$\text{kg}^{-1}\cdot\text{m}^{-1}\cdot\text{s}^2\cdot\text{A}^2$	$\text{m}\cdot\text{H}^{-1}$	1/[Permeability]
Relative permeability, magnetic	1	Dimensionless	[Permeability]/[Permeability of vacuum]
Susceptibility, magnetic	1	Dimensionless	[Relative permeability] - 1
Characteristic impedance	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	$\text{V}\cdot\text{A}^{-1}$, Ω , ohm	$\sqrt{([\text{Mag.Permeability}]/[\text{El.Permittivity}]}$
Electric Dielectric strength rigidity	$\text{kg}\cdot\text{m}\cdot\text{s}^{-3}\cdot\text{A}^{-1}$	$\text{V}\cdot\text{m}^{-1}$	$[\Delta\text{Potential}]/[\text{Distance}]$
Verdet constant	$\text{kg}^{-1}\cdot\text{m}^{-1}\cdot\text{s}^2\cdot\text{A}^1$	$\text{rad}\cdot\text{m}^{-1}\cdot\text{T}^{-1}$	$([\text{Angle}]/[\text{Length}])/[\text{Magnetic flux density}]$
Work function	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-2}$	J, eV	[Energy] needed to remove an electron
Thermoelectric power Thermopower	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-1}\cdot\text{K}^{-1}$	$\text{V}\cdot\text{K}^{-1}$	$[\Delta\text{Potential}]/[\Delta\text{Temperature}]$
Seebeck coefficient	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-1}\cdot\text{K}^{-1}$	$\text{V}\cdot\text{K}^{-1}$	$[\Delta\text{Potential}]/[\Delta\text{Temperature}]$
Thomson coefficient	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-1}\cdot\text{K}^{-1}$	$\text{W}\cdot\text{K}^{-1}\cdot\text{A}^{-1}$	[Heat flux]/([Temperature]*[Current])
Peltier coefficient	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-1}$	$\text{W}\cdot\text{A}^{-1}$, V	[Heat flux]/[Current]
Piezoelectric coefficient	$\text{kg}\cdot\text{m}\cdot\text{s}^{-3}\cdot\text{A}^{-1}$	$\text{V}\cdot\text{m}^{-1}$	[El.field strength]/([ΔLength]/[Length])
Electrostriction coefficient	$\text{kg}^{-2}\cdot\text{m}^{-2}\cdot\text{s}^6\cdot\text{A}^2$	$\text{m}^2\cdot\text{V}^{-2}$	$([\Delta\text{Volume}]/[\text{Volume}])/[\text{El.field strength}]^2$
g-factor of a particle	1	Dimensionless	[Mag.moment]/([Spin].[Bohr magneton])

Properties of electric/magnetic devices and circuit components

Bandwidth	s^{-1}	Hz	[ΔFrequency]
Voltage Electromotive force (emf)	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-1}$	V	[ΔPotential]
Current, electric	A	A	ampere . [Charge]/[Time]
Magnetomotive force (mmf)	A		[Current]*[Number of turns]
Impedance, of a circuit	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	Ω	ohm
Admittance, of a circuit	$\text{kg}^{-1}\cdot\text{m}^{-2}\cdot\text{s}^3\cdot\text{A}^2$	S	siemens . 1/[Circuit impedance]
Resistance	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	$\text{V}\cdot\text{A}^{-1}$, Ω (ohm)	[ΔPotential]/[Current]
Conductance	$\text{kg}^{-1}\cdot\text{m}^{-2}\cdot\text{s}^3\cdot\text{A}^2$	$\text{A}\cdot\text{V}^{-1}$, S (siemens)	1/[Resistance]
Capacitance	$\text{kg}^{-1}\cdot\text{m}^{-2}\cdot\text{s}^4\cdot\text{A}^2$	$\text{C}\cdot\text{V}^{-1}$, F	farad . [Charge]/[ΔPotential]
Reactance, capacitive	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	Ω (ohm)	1/(i[Angular frequency].[Capacitance])
Susceptance, capacitive	$\text{kg}^{-1}\cdot\text{m}^{-2}\cdot\text{s}^3\cdot\text{A}^2$	S (siemens)	1/[Reactance]
Inductance Mutual inductance	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-2}\cdot\text{A}^{-2}$	$\text{V}\cdot\text{s}\cdot\text{A}^{-1}$, $\text{Wb}\cdot\text{A}^{-1}$, H	henry . [ΔPotential]/[dCurrent/dt] or [Magnetic flux]/[Current]
Impedance, inductive	$\text{kg}\cdot\text{m}^2\cdot\text{s}^{-3}\cdot\text{A}^{-2}$	Ω (ohm)	i[Angular frequency].[Inductance]
Admittance, inductive	$\text{kg}^{-1}\cdot\text{m}^{-2}\cdot\text{s}^3\cdot\text{A}^2$	S (siemens)	1/[Inductive impedance]
Number of turns	1		Applicable to coils, transformers, etc
Current noise, variance n_j^2	$\text{s}\cdot\text{A}^2$	A^2/Hz	[Current] ² /[Bandwidth]
Voltage noise, variance n_v^2	$\text{kg}^2\cdot\text{m}^4\cdot\text{s}^{-5}\cdot\text{A}^{-2}$	V^2/Hz	[Voltage] ² /[Bandwidth]

Chemistry, physical chemistry, atomic and molecular physics

Concentration Molar density Molarity	$\text{m}^{-3}\cdot\text{mol}$		[Quantity]/[Volume]. Same as Density of substance
--	--------------------------------	--	--

Molality	$\text{kg}^{-1}.\text{mol}$	mol/kg	[Quantity]/[Mass]
Katalytic activity Molar production rate	$\text{mol}.\text{s}^{-1}$	katal	[Quantity]/[Time]
Molar mass	$\text{kg}.\text{mol}^{-1}$		[Mass]/[Quantity]
Molar charge	$\text{s}.\text{A}.\text{mol}^{-1}$	$\text{C}.\text{mol}^{-1}$	[Charge]/[Quantity]
Molecular ionic quantum charge	1	Dimensionless	[Charge of a molecule or ion]/[Elementary charge quantum]
Ionic strength Ionic force	$\text{m}^{-3}.\text{mol}$		$\text{Sum}([\text{Conc.}] \cdot [\text{Ionic quantum charge}]^2)$
Ion mobility	$\text{kg}^{-1}.\text{m}^{-1}.\text{s}^2.\text{A}$	$\text{m}^2.\text{s}^{-1}.\text{V}^{-1}$	[Velocity]/[Electric field strength]
Drift speed	$\text{m}.\text{s}^{-1}$		Steady-state speed of ions in electric field
Fugacity	$\text{kg}.\text{m}^{-1}.\text{s}^{-2}$	Pa	Effective pressure in real gases
Osmotic pressure	$\text{kg}.\text{m}^{-1}.\text{s}^{-2}$	Pa	
Thermodynamic force	$\text{kg}.\text{m}.\text{s}^{-2}.\text{mol}^{-1}$	N/mol	$[\Delta\text{Chemical potential}]/[\text{Distance}]$

Chemico-physical properties of elements

Atomic number	1	Dimensionless	Number of protons in an atomic nucleus
Atomic weight Relative atomic mass	au	atomic units	Average over a typical isotopic composition
Mass number of an isotope	1	Dimensionless	Number of protons+neutrons in the isotope nuclide
Electronegativity, Pauling χ	1	Dimensionless	Relative tendency of an atom to attract electrons; $\chi(\text{H})=2.20$.
Electron affinity (always molar)	$\text{kg}.\text{m}^2.\text{s}^{-2}.\text{mol}^{-1}$	$\text{J}.\text{mol}^{-1}$	Energy released when binding an electron

Chemico-physical properties of matter

Ionization energy, molar	$\text{kg}.\text{m}^2.\text{s}^{-2}.\text{mol}^{-1}$	$\text{J}.\text{mol}^{-1}$	Energy to ionize a molecule/atom
Volume, molar	$\text{m}^3.\text{mol}^{-1}$		[Volume]/[Quantity]
Heat of fusion evaporation, molar	$\text{kg}.\text{m}^2.\text{s}^{-2}.\text{mol}^{-1}$	$\text{J}.\text{mol}^{-1}$	[Energy]/[Quantity]
Chemical potential, molar	$\text{kg}.\text{m}^2.\text{s}^{-2}.\text{mol}^{-1}$	$\text{J}.\text{mol}^{-1}$	$[\Delta\text{InternalEnergy}]/[\Delta\text{Quantity}]$
Solubility, molar	$\text{m}^{-3}.\text{mol}$		[Quantity]/[Volume]
Reduction Redox potential	$\text{kg}.\text{m}^2.\text{s}^{-3}.\text{A}^{-1}$	V (volt)	
Conductivity, molar	$\text{kg}^{-1}.\text{s}^3.\text{A}^2.\text{mol}^{-1}$	$\text{S}.\text{m}^2.\text{mol}^{-1}$	[El.conductivity]/[Concentration]
Relaxivity, molar	$\text{s}^{-1}.\text{mol}^{-1}$		[Relaxation rate]/[Concentration]
Ebullioscopic constant	$\text{kg}.\text{mol}^{-1}.\text{K}$	K/(mol/kg)	$[\Delta\text{Temperature}]/[\text{Molality}]$
Cryoscopic constant	$\text{kg}.\text{mol}^{-1}.\text{K}$	K/(mol/kg)	$[\Delta\text{Temperature}]/[\text{Molality}]$
Compression factor of a real gas	1	Dimensionless	$pV/(nRT)$. For ideal gas equals 1; temperature dependent
van der Waals constant: a	$\text{kg}.\text{m}^5.\text{s}^{-2}.\text{mol}^{-2}$	$\text{Pa}.\text{m}^6$	a in $(p+a/V^2)(V-b)=RT$, where V is molar volume
van der Waals constant: b	$\text{m}^3.\text{mol}^{-1}$		b in $(p+a/V^2)(V-b)=RT$, where V is molar volume
Virial coefficient: second	$\text{m}^3.\text{mol}^{-1}$		B in $pV/(nRT)=1+B(n/V)+C(n/V)^2+D(n/V)^3+\dots$
Virial coefficient: third	$\text{m}^6.\text{mol}^{-2}$		C in $pV/(nRT)=1+B(n/V)+C(n/V)^2+D(n/V)^3+\dots$
Virial coefficient: fourth	$\text{m}^9.\text{mol}^{-3}$		C in $pV/(nRT)=1+B(n/V)+C(n/V)^2+D(n/V)^3+\dots$

Gravitation, Astronomy, Cosmology

Gravitational field intensity Gravity	$\text{m}.\text{s}^{-2}$		[Force]/[Mass], Same as acceleration
Gravitational field potential	$\text{m}^2.\text{s}^{-2}$		[Energy]/[Mass]
Gravitational constant G	$\text{kg}^{-1}.\text{m}^3.\text{s}^{-2}$		[Force]*[Distance] ² /[Mass] ² . Appears in Newton's equation
Mean motion	s^{-1}		Of a body on a Kepler orbit; $\sqrt{G(M_1+M_2)/r^3}$
Mean anomaly	1	Dimensionless	Of a body on a Kepler orbit; $t \cdot \sqrt{G(M_1+M_2)/r^3}$
Star magnitude (astronomy)	1	Dimensionless	$m-m' = -2.5 \log(S/S')$. S, S' are luminous fluxes of two stars
Cosmological constant Λ	m^{-2}		Appears in Einstein's equation
Cosmological expansion rate	s^{-1}	km/s/Mpc	[Velocity]/[Distance]. Mpc stands for Megaparsec

Optics

Albedo, of a surface	1	Dimensionless	[Reflected elmag power]/[Incident elmag power]
Convergence	m^{-1}	dioptry	dioptry
Luminosity Luminous intensity	cd	cd	candle or lumen/sr
Luminous flux Luminous power	cd.sr	lm	lumen. [Luminosity]*[Solid angle]
Luminance	$\text{cd}.\text{m}^{-2}$		[Luminosity]/[Area]
Luminous energy	cd.sr.s	lm.s	[Luminous flux]*[Time]. Also known as talbot
Illuminance	$\text{cd}.\text{sr}.\text{m}^{-2}$	$\text{lm}.\text{m}^{-2}$, lx	lux. [Luminous flux]/[Area]
Luminous emittance	$\text{cd}.\text{sr}.\text{m}^{-2}$	$\text{lm}.\text{m}^{-2}$, lx	lux. Same as illuminance, but for sources
Luminous efficacy	$\text{cd}.\text{sr}.\text{kg}^{-1}.\text{m}^{-1}.\text{s}^3$	lm/W	[Luminous flux]/[Power]
Luminous efficiency Luminous coefficient	1	Dimensionless	[Luminous efficacy]/[683 lm/W]

Irradiance	kg.s ⁻³	W.m ⁻²	[Power]/[Area]. For all kinds of energy deposition
Radiance	kg.s ⁻³ .sr ⁻¹	W.m ⁻² .sr ⁻¹	([Power]/[Area])/[Solid angle]
Optical properties of matter			
Extinction coefficient	m ⁻¹		
Refractive index	1	Dimensionless	Light speeds ratio (in medium)/(in vacuum)
Specific refractivity	m ³ .kg ⁻¹		$[(r^2-1)/(r^2+2)]$ /[Specific density], where r is refractive index
Molar refractivity	m ³ .mol ⁻¹		$[(r^2-1)/(r^2+2)]$ [Concentration]
Dispersivity quotient	m ⁻¹		$[\Delta\text{Refractive index}]/[\Delta\text{Wavelength}]$
Dispersive power	1	Dimensionless	Ratio of differences of refractive indices
Constringence Abbé number V-number	1	Dimensionless	$V_D = (n_D-1)/(n_F-n_C)$
Radiation and radioactivity			
Radioactivity Activity	s ⁻¹	Bq	becquerel . [Counts]/[Time]
Irradiance	kg.s ⁻³	W.m ⁻²	[Power]/[Area]. For all kinds of energy deposition
Absorbed dose	m ² .s ⁻²	J.kg ⁻¹ , Gy	gray . [Energy]/[Mass]
Absorbed dose rate	m ² .s ⁻³	Gy.s ⁻¹	[Absorbed dose]/[Time]
Absorbed dose equivalent	m ² .s ⁻²	J.kg ⁻¹ , Sv	sievert . [const]. [Energy]/[Mass]
Exposure	kg ⁻¹ .s.A	C.kg ⁻¹	[Charge]/[Mass]. For ionising radiations
Radiation properties of matter			
Half life	s		Of a radioisotope
Radiation power	m ² .s ⁻³	W/kg	[Power]/[Mass]. Heat generated by a radioisotope
Radiation power, molar	kg.m ² .s ⁻³ .mol ⁻¹	W/mol	[Power]/[Quantity]. Heat generated by a radioisotope
Informatics			
Information	bit ⁻¹	bit	bit ; the elementary information quantum
Baud rate Information flux	bits.s ⁻¹	Baud	baud . [Information]/[Time]
Economy and finance			
Transactions count	1	Dimensionless	All kinds of counts
Interest	1	%	$[\Delta\text{Wealth}]/[\text{Wealth}]$. Usually expressed as percentage
Wealth Asset	cur	currency	Currencies like \$, EUR, Yuan, ... are different units
Debt Liability	cur	currency	Usually intended as negative wealth
Value Price	cur	currency	Prefixes: K ..thousands, M ..millions, B ..billions
Transaction value Sale Purchase	cur	currency	Often used: mean and total values
Time period	s	year, quarter, month	Abbrevs: mrq .. most recent quarter, ttm .. trailing twelve months
Fiscal year Calendar year	s	year	Abbrevs: lfy .. last fiscal year, yoy .. year over year
Transactions rate Activity	s ⁻¹	1/year	[Transactions]/[Time period]
Transactions volume Sales flow	cur.s ⁻¹		[Value]/[Time period]. For example \$/day or Eur/year
Velocity / circulation of money	s ⁻¹	1/year	[Transactions]/[Time period]
Interest rate	s ⁻¹	%/year	[Interest]/[Time period]
Return on asset / equity	s ⁻¹	%/year	$([\Delta\text{Value}]/[\text{Value}])/[\text{Time period}]$
Cash flow Flow (generic)	cur.s ⁻¹	currency/year	[Value]/[ΔTime]. Mathematically, time derivative
Earnings Income rate	cur.s ⁻¹	currency/year	[Value]/[Time period]
GDP Gross domestic product	cur.s ⁻¹	currency/year	[Earnings]. Usually refered to nations/states/admin.regions
Debt/GDP ratio	s	year	[Debt]/[Earnings]. Independent of currency / population size
P/E Price/Earnings ratio	s	year	[Value]/[Earnings]. Used to assess an asset/company
Bond duration	s	year	In general, the duration of a fixed cash flow

Notes

Purpose

Physical (or rather metrological) dimensions are often bewildering, even though the [international SI system of units](#) has simplified things a lot, compared to early 20th century and before. The main purpose of this page is to provide a **fast, handy reference** to the dimension you might need at the spur of a moment. Another, less evident, purpose is to **stimulate curiosity** and the desire to study Metrology and Dimensional Analysis.

Formats and editorial comments

- **Bold magenta symbols** in the **Alternatives** column indicate commonly used quantities, mostly defined by the SI system.
- **Square brackets** convert the quantity they enclose into its *dimension*.
- Abbreviations **EI** and **Mag** stand for **Electric** and **Magnetic**, respectively.
- [Quantity] stands for [Quantity of substance] and its dimension is **mol**.
- Names of units are always written with small first letter, even when derived from names of persons (for example 1 newton).

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Disclaimer:

Since errors do happen, and also because not all metrological conventions are agreed upon and shared by everybody, the Editor of this page declines any responsibility for any damages that might result from its content, directly or indirectly. In other words, if you crash a spacecraft because some of your engineers used *meters* and others used *feet*, do not pretend that I should pay for it :-)

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- 15. For more, see [References on Systems of Units of Measurements](#)

Links

- [Dimensional analysis](#) on Wikipedia.
- [BIPM](#). The home page of the SI System of Units.
- [NIST Units of Measurements](#) page.
- [NIST Units Bibliography](#) to official on-line publications about the SI system.
- [Unit Converters](#) list.
- [SI Units](#). A related resource on this website.
- [Constants of physics and mathematics](#). A related resource on this website.
- [Math Constants](#). A related resource on this website.
- [SI Dimensions of Physical Quantities](#). A related resource on this website (sorted alphabetically).
- [Dimensions of Physical Quantities by Category](#). Link to this resource. You can also cite its **DOI link: 10.3247/SL1Phys06.004**.
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